

KINETIC ANALYSIS OF OUABAIN- $K^+$  AND  $Na^+$  INTERACTION ON A  
 $Na^+, K^+$ -DEPENDENT ADENOSINETRIPHOSPHATASE FROM CARDIAC TISSUE\*

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Antagonism by  $K^+$  of ouabain inhibition of  $Na^+, K^+$ -dependent adenosinetriphosphatases ( $Na^+, K^+$ -ATPase), reported earlier by Dunham and Glynn (1961), Post and Albright (1961) and Auditore (1964), suggests that ouabain inhibits the enzyme activity by a displacement of  $K^+$  from a  $K^+$ -binding site (Skou, 1964). On the other hand, Schatzmann (1965) observed a non-competitive interaction between ouabain and  $K^+$  or  $Na^+$  on a red cell ghost ATPase system.

The kinetics of the  $Na^+, K^+$ -ATPase as well as studies of partial reactions of the enzyme system have not been extensively pursued due, in part, to difficulty in purifying the enzyme. The present report is concerned with the kinetics of ouabain inhibition of a highly specific  $Na^+, K^+$ -ATPase from cardiac tissue (Matsui and Schwartz, 1966). The non-competitive nature of ouabain- $K^+$  interaction is described, stressing the dependency of ouabain-induced inhibition upon the  $Na^+/K^+$  ratio.

#### METHODS

Enzyme: The preparation of a highly specific  $Na^+, K^+$ -ATPase from

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calf heart has been recently described (Matsui and Schwartz, 1966). More than 95% of the total ATPase activity of the enzyme was  $\text{Na}^+, \text{K}^+$ -dependent and ouabain-sensitive; the specific activity varied from 20 to 30  $\mu\text{moles Pi/mg protein/hr.}$

Assay of ATPase: The reaction mixture, in a total volume of 1 ml, contained 50 - 100  $\mu\text{g}$  enzyme protein, 2 mM ATP (Tris-salt), 5 mM  $\text{MgCl}_2$ , 1 mM ethylenediaminetetraacetate, 50 mM Tris-HCl (pH 7.4), and various concentrations of NaCl and KCl in the presence or absence of ouabain. Incubation was carried out at  $37^\circ$  for 10 min. After addition of ice-cold trichloroacetic acid, the liberated orthophosphate was determined by the method of Martin and Doty (1949). Protein was determined by the method of Lowry et al. (1951).

All results are expressed as  $\text{Na}^+, \text{K}^+$ -dependent ATPase activity which was calculated from the activity in the presence of  $\text{Na}^+$  and  $\text{K}^+$ , subtracting the basal  $\text{Mg}^{2+}$ -dependent activity.

## RESULTS AND DISCUSSION

At a constant level of  $\text{Na}^+$  concentration, the percent inhibition produced by ouabain decreased with increasing concentrations of  $\text{K}^+$ ; at a constant level of  $\text{K}^+$ , the percent inhibition by ouabain increased with increasing concentrations of  $\text{Na}^+$ . This finding, suggesting that  $\text{Na}^+$  as well as  $\text{K}^+$  affects ouabain inhibition, is in agreement with the work of Schatzmann (1965) and in contrast to that of Post and Albright (1961). The kinetics of the system with the above conditions, however, are difficult to interpret. This is probably due to the complicating presence of  $\text{K}^+$  inhibition at the  $\text{Na}^+$ -site and of  $\text{Na}^+$  inhibition at the  $\text{K}^+$ -site (Green and Taylor, 1964).

When the ratio of  $\text{Na}^+$  and  $\text{K}^+$  was kept constant in order to cancel the inhibitory interaction between  $\text{Na}^+$  and  $\text{K}^{+*}$ , however, a

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double-reciprocal plot of ATPase activity and  $K^+$  (i.e.  $Na^+$ ), yielded a linear relationship except at very low concentrations of  $K^+$  ( $\leq 0.2$  mM  $K^+$ ). The inhibition by ouabain at a constant  $Na^+/K^+$  ratio was analyzed by the method of Lineweaver and Burk (1934). As shown in Fig. 1, maximal velocities in the presence and absence of two concentrations of ouabain were dissimilar, and, moreover, each line met at the same intercept on the abscissa, indicating a non-competitive type of inhibition between ouabain and  $K^+$  (or  $Na^+$ ).

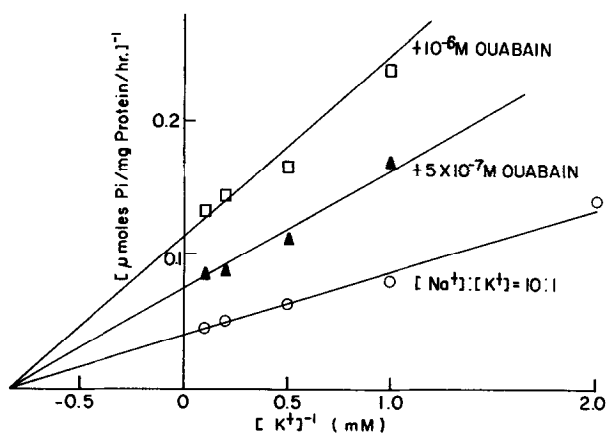


Fig. 1. Lineweaver-Burk analysis of ouabain inhibition at a constant  $Na^+/K^+$  ratio.  $Na^+, K^+$ -ATPase activity was determined at various  $K^+$  concentrations with a constant  $Na^+/K^+$  ratio of 10 in the presence and absence of  $5 \times 10^{-7}$  M and  $10^{-6}$  M ouabain;  $1/v$  was plotted against  $1/K^+$  ( $=1/Na^+ \times 10$ ).

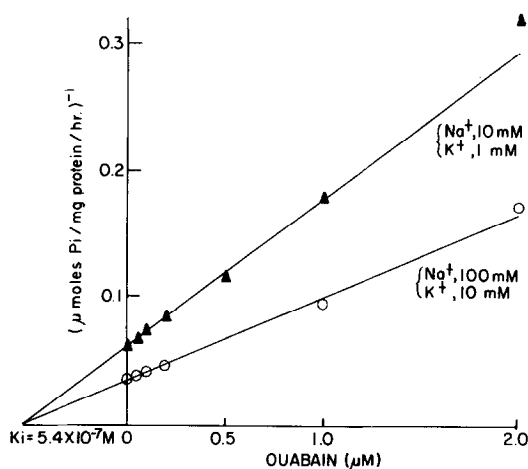


Fig. 2. Dixon plot of ouabain inhibition at a constant  $Na^+/K^+$  ratio.

This non-competitive type of inhibition was also demonstrated by the method of Dixon (1953). The plot of  $1/v$  against ouabain concentration yielded a linear relationship, and the lines corresponding to different  $K^+$  concentrations with the same  $Na^+/K^+$  ratio

met at the same intercept on the abscissa (Fig. 2). At a  $\text{Na}^+/\text{K}^+$  ratio of 10, the  $K_i$  for ouabain obtained from the Dixon method (Fig. 2) was  $5.3 \times 10^{-7} \text{M}$ , and was the same as that calculated from the Lineweaver-Burk method (Fig. 1).

When the Dixon plot was constructed at different concentrations of  $\text{K}^+$  and a constant level of  $\text{Na}^+$  (Fig. 3a), the lines met approximately at one point, suggesting competitive kinetics. On the other hand, when  $\text{Na}^+$  was varied while maintaining the  $\text{K}^+$  concentration constant (Fig. 3b), the plots described neither competitive nor non-competitive kinetics (Dixon, 1953). However, since the ouabain- $\text{K}^+$  or  $\text{Na}^+$  interaction is clearly of a non-competitive nature as shown in Figs. 1 and 2, there must be another interpretation for the results shown in Figs. 3a and b. It is proposed that the  $K_i$  for ouabain is dependent upon the  $\text{Na}^+/\text{K}^+$  ratio. In Fig. 3, therefore, each line represents merely the different  $K_i$ 's for ouabain, corresponding to a given ratio of  $\text{Na}^+/\text{K}^+$  and the overall "family" of lines has no validity in determining the type of kinetics involved. Increasing the  $\text{K}^+$  at a constant  $\text{Na}^+$  concentration (i.e. decreasing the  $\text{Na}^+/\text{K}^+$  ratio) caused an increase in the  $K_i$  (intercept on abscissa) (Fig. 3a), while increasing the  $\text{Na}^+$  at a constant  $\text{K}^+$  concentration (i.e. increasing the  $\text{Na}^+/\text{K}^+$  ratio) decreased the  $K_i$  (Fig. 3b). The relationship between  $K_i$  and  $\text{Na}^+/\text{K}^+$  ratio is graphically displayed in Figure 4 and may be mathematically expressed as:

$$K_i = 3.5 \times 10^{-6} \times (\text{Na}^+/\text{K}^+)^{-0.82}$$

This relationship obtains in the range of  $\text{Na}^+/\text{K}^+$  ratio from 1 to 100.

From the results presented here, it is concluded that inhibition by ouabain is due neither to direct nor indirect displacement of  $\text{K}^+$  at the  $\text{K}^+$ -site. Furthermore, the binding site of ouabain appears to be different from the  $\text{K}^+$ -site. Although ouabain inhibi-

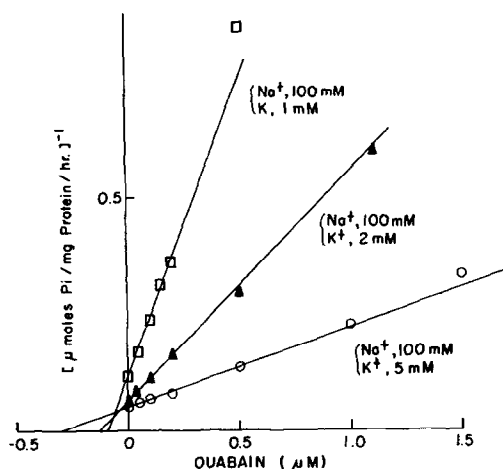


Fig. 3a. Dixon plot of ouabain inhibition at different  $\text{Na}^+/\text{K}^+$  ratios. Varying  $\text{K}^+$  with a constant  $\text{Na}^+$  concentration.

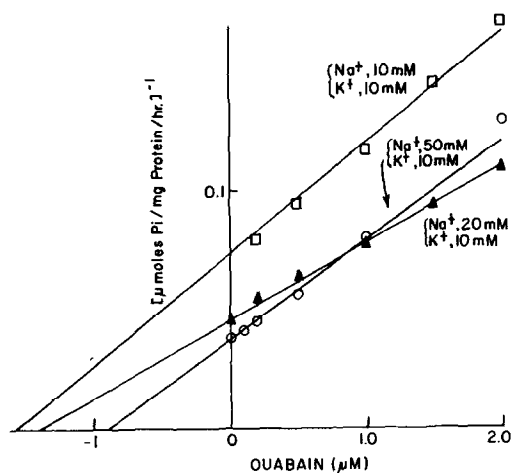
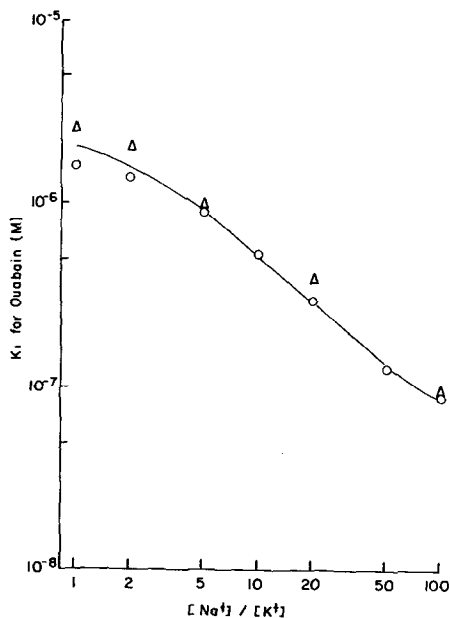


Fig. 3b. Dixon plot of ouabain inhibition at different  $\text{Na}^+/\text{K}^+$  ratios. Varying  $\text{Na}^+$  with a constant  $\text{K}^+$  concentration.

Fig. 4. Relationship between  $K_i$  for ouabain and  $\text{Na}^+/\text{K}^+$  ratio.  $K_i$  obtained was plotted against  $\text{Na}^+/\text{K}^+$  ratios on a full logarithmic scale.  $\circ, \Delta$  represent different enzyme preparations.



tion is not competitive with  $K^+$ , the dependency of the  $K_i$  for ouabain on the  $Na^+/K^+$  ratio (Fig. 3a) could explain the antagonistic phenomenon between ouabain and  $K^+$  reported earlier (Dunham and Glynn, 1961; Post and Albright, 1961; Auditore, 1964). These results are in accordance with Schatzmann's observations (1965), dealing with the interaction of  $Na^+$  on ouabain inhibition. In this connection, inhibition by ouabain of a  $K^+$ -dependent p-nitrophenylphosphatase (Fujita et al., 1966) is of interest, for the  $K_i$  ( $3 \times 10^{-6}M$ ) in the absence of  $Na^+$  is similar to the upper limit of the  $K_i$  on the  $Na^+, K^+$ -ATPase at low  $Na^+/K^+$  ratios (Fig. 4).

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